

CHEMISTRY 103
SUMMER 2019
Syllabus

Day	Topic	Pages in OpenStax
May 28	I. Matter & Measurement (Chapter 1, Appendices B and C) A. Domain and methods of chemistry B. Calculations: units, digits and uncertainty	1-29 29-51, 1213-1222
May 29	II. Atomic Structure (Chapters 2 and 6) A. Early chemical laws B. Modern atomic structure	67-78 78-86
May 30	C. Atomic mass, Avogadro's number and the mole D. Percent composition and empirical formula	307-310 311-318
May 31	III. Electronic Structure and the Periodic Table (Chapter 3) A. Electromagnetic radiation and quantization B. The Bohr atom and atomic line spectra	115-128 128-134
June 3	C. Quantum mechanics and hydrogen-like orbitals D. Periodic table and electron filling in atoms E. Periodic trends	134-148 148-157 157-169
June 4	IV. Ionic Bonding (Chapters 3 and 4) A. Electronegativity and bond polarity B. Ions and ionic bonding C. Ionic nomenclature and polyatomic ions	199-203 169-172, 193-197 172-177, 203-207
June 5	V. Covalent Bonding (Chapter 4) A. Molecules and covalent bonding B. Covalent nomenclature C. Lewis dot structures, resonance D. Valence shell electron pair repulsion model	197-199, 236-239 207-210 210-217, 220-225 225-236
June 6	VI. Chemical Reaction Stoichiometry (Chapters 6 and 7) A. Chemical equations B. Stoichiometric calculations C. Solution stoichiometry	341-348 361-371 318-326
June 10	VII. Chemical Reaction Types (Chapters 7 and 11) A. Electrolytes, ions and net ionic equations B. Precipitation and acid-base reactions	603-606, 346-348 348-355
June 11	C. Oxidation-reduction reactions D. Titrations and gravimetry	355-361 371-378

Day	Topic	Pages in OpenStax
	VIII. Gases (Chapter 8)	
June 12	A. Gas pressure and the kinetic molecular theory	395-405, 435-440
	B. Diffusion and effusion	430-435
	C. Gas laws	405-427
June 13	D. Real gases	441-444
	IX. Thermochemistry (Chapters 9, 12, and 13)	
June 14	A. Heat, work, energy, enthalpy, and calorimetry	459-490
	B. Standard enthalpies of formation and Hess's Law	491-497
	C. Bond dissociation energies	497-502
June 17	D. Entropy and free energy	653-670, 706-708
	X. Liquids and Solids (Chapter 10)	
June 18	A. Intermolecular forces	519-532
	B. Liquids	532-549
	C. Solids	556-568
June 19	D. Phase diagrams	549-556
	XI. Solutions (Chapters 6 and 11)	
June 20	A. Concentration measurements and solubility	326-331, 597-603
	B. Henry's and Raoult's laws	606-620
	C. Boiling-point elevation and freezing-point depression	620-627
	D. Osmotic pressure	627-633
	XII. Chemical Kinetics (Chapter 17)	
June 24	A. Reaction rates	895-904
	B. Rate laws and reaction order	904-911
	C. First-order reactions	911-914, 918-920
	D. Kinetics and mechanism	921-933
	E. Catalysts and temperature effects	933-938
June 25	XIII. Chemical Equilibrium (Chapter 13)	
	A. Equilibrium and equilibrium constant	679-692, 695-706
	B. Le Châtelier's Principle	692-695
	XIV. Acids and Bases (Chapter 14)	
June 26	A. Nature of acids and bases	731-736
	B. pH scale	736-741
	C. Equilibrium calculations for weak acids and bases	742-758
June 27	D. Acid-base properties of salts	760-764
	E. Common ion effect and buffers	771-779

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Text: E. J. Neth, P. Flowers, K. Theopold, R. Langley, W. R. Robinson, *Chemistry: Atoms First*, OpenStax: Houston, TX, ISBN: 9781938168154 (2016). <https://openstax.org/details/books/chemistry-atoms-first>

Instructor:	<u>Office</u>	<u>Phone</u>	<u>Office Hours</u>	<u>E-Mail</u>
Robert Pike	ISC 2043	221-2555	Daily, 1:30–2:30 PM	rdpike@wm.edu

Course Goals: This course is intended for science concentrators and pre-medical students. It introduces the student to the nature of atoms and molecules, stoichiometry, states of matter, solutions, reactions, kinetics and equilibrium.

Lectures: Monday through Friday ISC 1127, 8:00–9:30 AM

Examinations: Each of the three exams covers about a third of the course material and contains (i) problems requiring numerical answers similar to the problems in the problem sets, (ii) short-answer questions, and (iii) multiple-choice questions. Your weakest exam will be counted only half as much as your other two exams.

Grading:	<u>Syllabus Topics</u>	<u>Chapters in OpenStax</u>	<u>Date</u>
34%* First Test	I – V	1 – 4	June 7 (Friday)
34%* Second Test	VI – X	6 – 10, 12	June 21 (Friday)
34%* Third Exam	XI – XIV	11, 13, 14, 17	June 28 (Friday)
15% Homework Sets	(coverage and due dates on reverse side)		

*The weakest of the three exams will count only half as much (17%) as the other two (34% each).

Problem Sets: Working problems is important for reinforcing the chemical principles emphasized in the lecture and text.

Homework Sets: There are graded ten homework set assignments. The coverage and availability dates of the homework sets are listed on the reverse side of this page. These homework sets are available through *Sapling Learning*. Each set is due by 11:59 PM on the day indicated. Each set will be automatically graded through *Sapling*. To help with the learning process, you get three tries to get correct answers for each problem. The homework set deadlines are firm; no homework sets will be accepted late. You may work in small groups; however each student is ultimately responsible for mastering the material for him/herself. Solutions to the assigned problems will be posted on *Sapling* after the homework set is due.

Practice Problems: There are numerous problems and exercises within and at the end of each text chapter. Many of these problems are very similar to the assigned problems in the homework sets. You should practice these problems if you are having difficulty with an assigned problem. A suggested list for each chapter is given on the reverse side of this page.

Sapling Homework Sets (graded)

Problem Set #	Lecture Units	Date Available	Date Due
1	I, IIA	May 27 th 8:00 AM	June 1 st 11:59 PM
2	IIB-D, IIIA-B	May 30 th 8:00 AM	June 1 st 11:59 PM
3	IIIC-E, IV	June 3 rd 8:00 AM	June 5 th 11:59 PM
4	V, VIA-B	June 5 th 8:00 AM	June 8 th 11:59 PM
5	VIC, VII	June 10 th 8:00 AM	June 12 th 11:59 PM
6	VII	June 12 th 8:00 AM	June 15 th 11:59 PM
7	IX, XA	June 17 th 8:00 AM	June 19 th 11:59 PM
8	XB-D, XI	June 19 th 8:00 AM	June 22 nd 11:59 PM
9	XII, XIII A	June 24 th 8:00 AM	June 26 th 11:59 PM
10	XIIIB, XIV	June 26 th 8:00 AM	June 28 th 11:59 PM

Additional Practice Problems (not graded)

Chapter	Problems
1	3, 9, 11, 13, 15, 17, 19, 23, 35, 37, 39, 45, 47, 49, 51, 53, 55, 59, 65, 71, 77, 81, 87, 89, 91, 93, 97
2	1, 3, 5, 7, 11, 17, 19, 25, 29, 37, 39, 41, 46, 45, 47, 49, 51, 53, 55, 57, 61
3	3, 5, 7, 9, 11, 15, 17, 19, 21, 23, 27, 33, 35, 37, 41, 45, 9, 53, 55, 57, 61, 67, 69, 71, 73, 75, 77, 79, 81, 83, 85, 93, 97, 99
4	3, 5, 7, 9, 13, 15, 21, 23, 25, 27, 29, 31, 40, 46, 48, 50, 52, 66, 70, 72, 79, 85(a-e), 89, 91, 95, 99
6	3, 8, 12, 14, 18, 22, 26, 28, 30, 32, 36, 38, 40, 42, 46, 48, 52, 54
7	3, 5, 7, 9, 11, 13, 17, 19, 21, 25, 29, 31, 33, 37, 39, 41, 43, 45, 47, 51, 57, 61, 63, 65, 71, 73, 75, 79, 81, 83, 87, 89, 93
8	5, 7, 15, 27, 29, 31, 33, 37, 39, 43, 45, 49, 51, 53, 55, 57, 61, 63, 65, 69, 75, 81, 85, 87, 89, 91, 95(a,b), 101, 103
9	7, 9, 11, 19, 21, 23, 25, 27, 31, 41, 49, 53, 55, 57, 59, 61, 63, 65, 67, 69, 71, 81, 88, 91, 92, 94, 100, 104
10	5, 7, 9, 11, 15, 21, 27, 35, 37, 39, 47, 51, 53, 55, 57, 65, 69, 73, 75, 77, 85
11	5, 9, 13, 15, 23, 25, 33, 35, 39, 41, 47, 49, 59, 65
12	3, 15, 17, 19, 21, 25, 31, 33, 37, 51
13	3, 5, 7, 9, 13, 15, 17, 33, 37, 39, 41, 45, 47, 49, 51, 53, 55, 65, 69, 73, 75, 77, 79, 81, 85, 87, 95
14	3, 5, 7, 9, 11, 19, 21, 25, 29, 33, 35, 47, 49, 61, 65, 67, 69(a-d), 71, 79(b-d), 87, 89, 91, 95, 97
17	3, 7, 13, 15, 19, 23, 25, 27, 29, 37, 45, 51, 53, 55, 63

How to access Sapling: Go to www.saplinglearning.com/login to log in or create an account.

- Under Enroll in a new course, you should see Courses at William & Mary. Click to expand this list and see courses arranged by subject. Click on a subject to see the terms that courses are available.
- Click on the term to expand the menu further (note that Semester 1 refers to the first course in a sequence and not necessarily the first term of the school year).
- Once the menus are fully expanded, you'll see links to specific courses. Click on the link for *The College of William and Mary - CHEM 103 - Summer19 - PIKE*.
- Review the system requirements and confirm that Flash is updated and enabled in your browser.