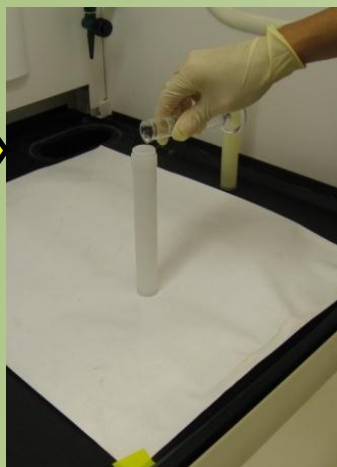


Overview of DR R's Research Interests

- Environmental analysis of heavy metals in sediments, fish, and other matrices
- Development of methodologies for the determination of methyl mercury in fish and other matrices
- Evaluation of new micro-columns as sample adsorbents and GC columns in lab-on-a-chip designs



Heavy Metals in Sediments and Fish



Several instruments are used for elemental determinations:



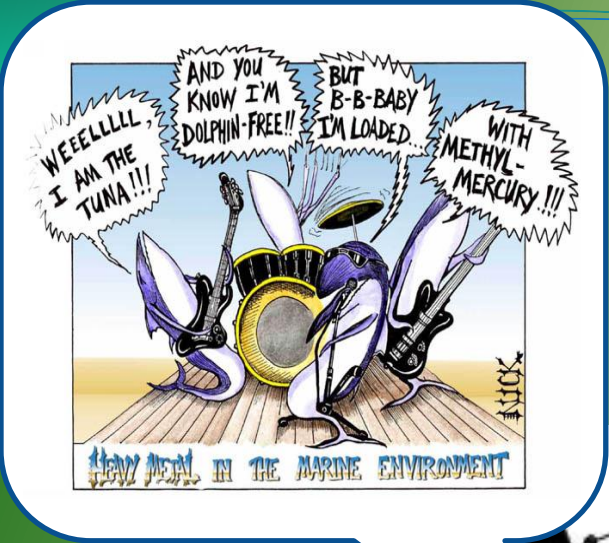
Graphite furnace atomic absorption spectrometry

Inductively coupled plasma spectroscopy

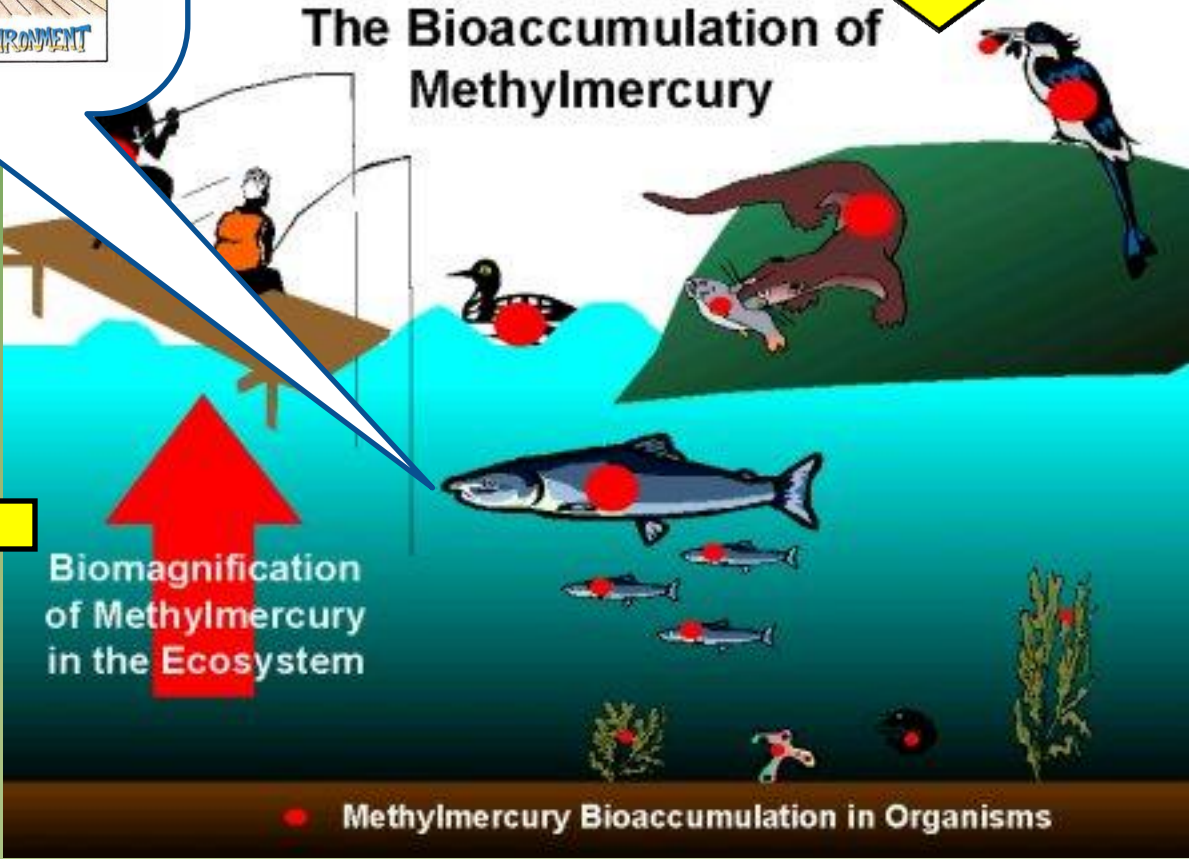


Automated mercury analyzer – based on atomic absorption from reduction of mercury via SnCl_2 (sub-ppb limits of detection)

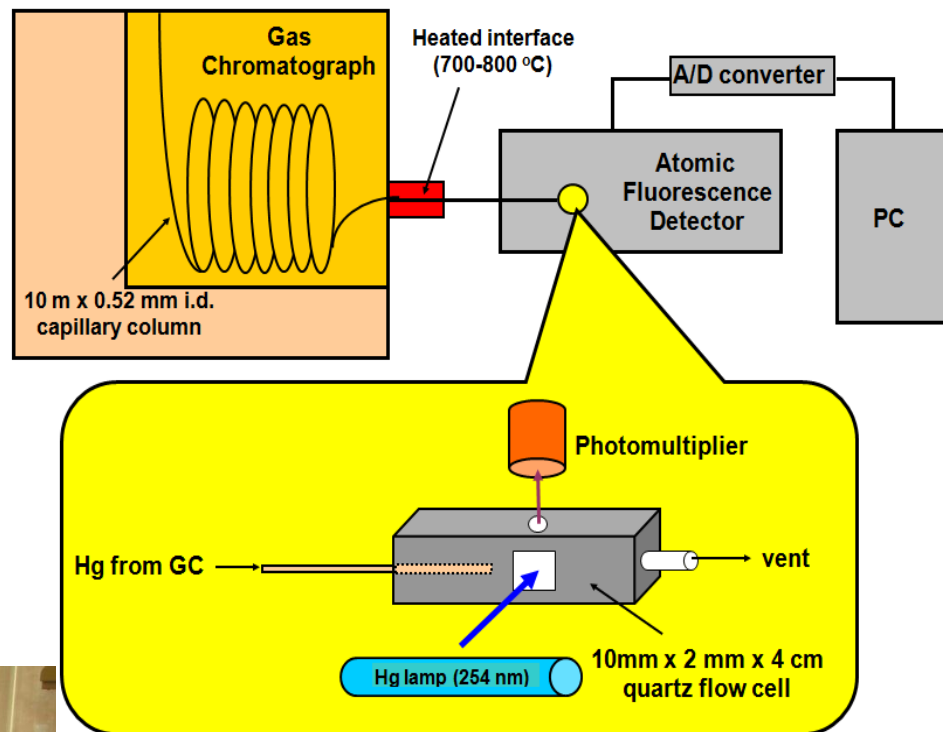


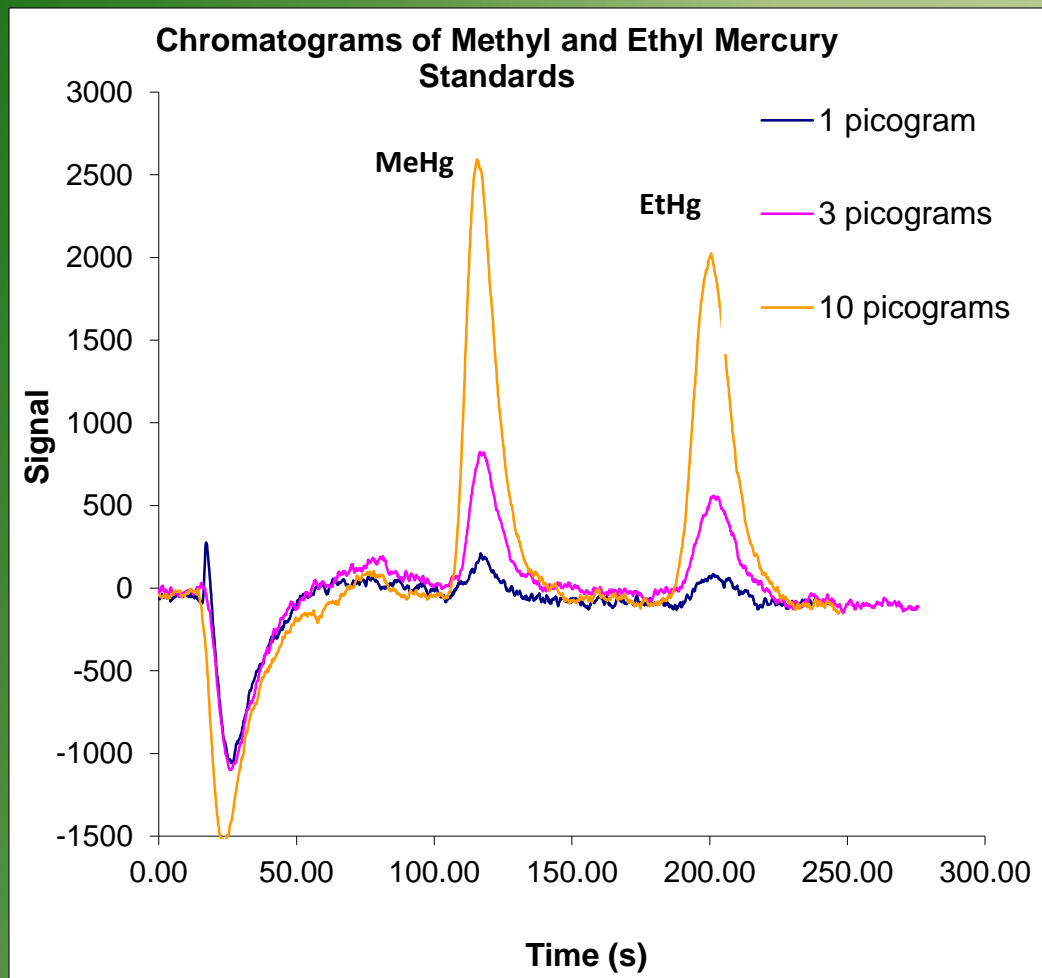


The Bioaccumulation of Methylmercury



Gas Chromatography- Atomic Fluorescence Spectroscopy (GC-AFS)



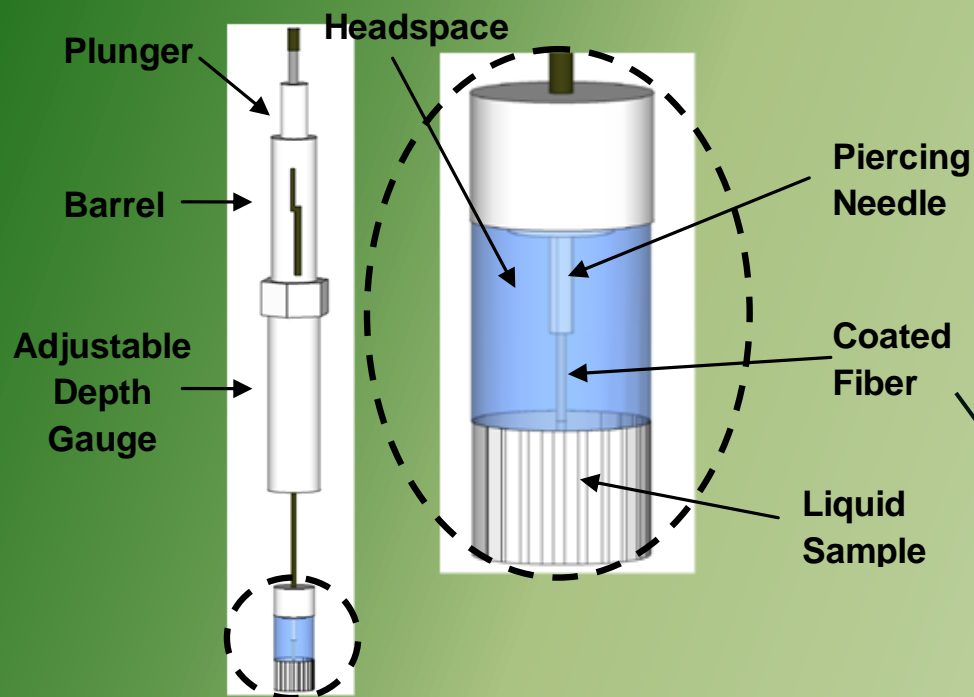


Current research is focused on simply extracting methylmercury as the chloride salt (MeHgCl) and using EtHgCl as an internal standard. Both of these can be separated using GC.

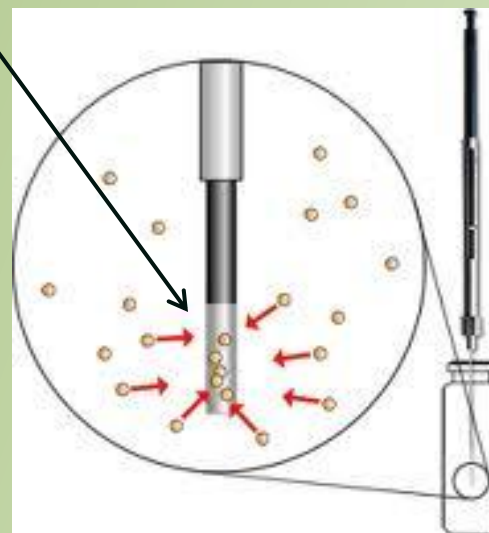


An extraction procedure is being developed for removal of MeHg from tissue by using saturated salt solutions and sonication methods.

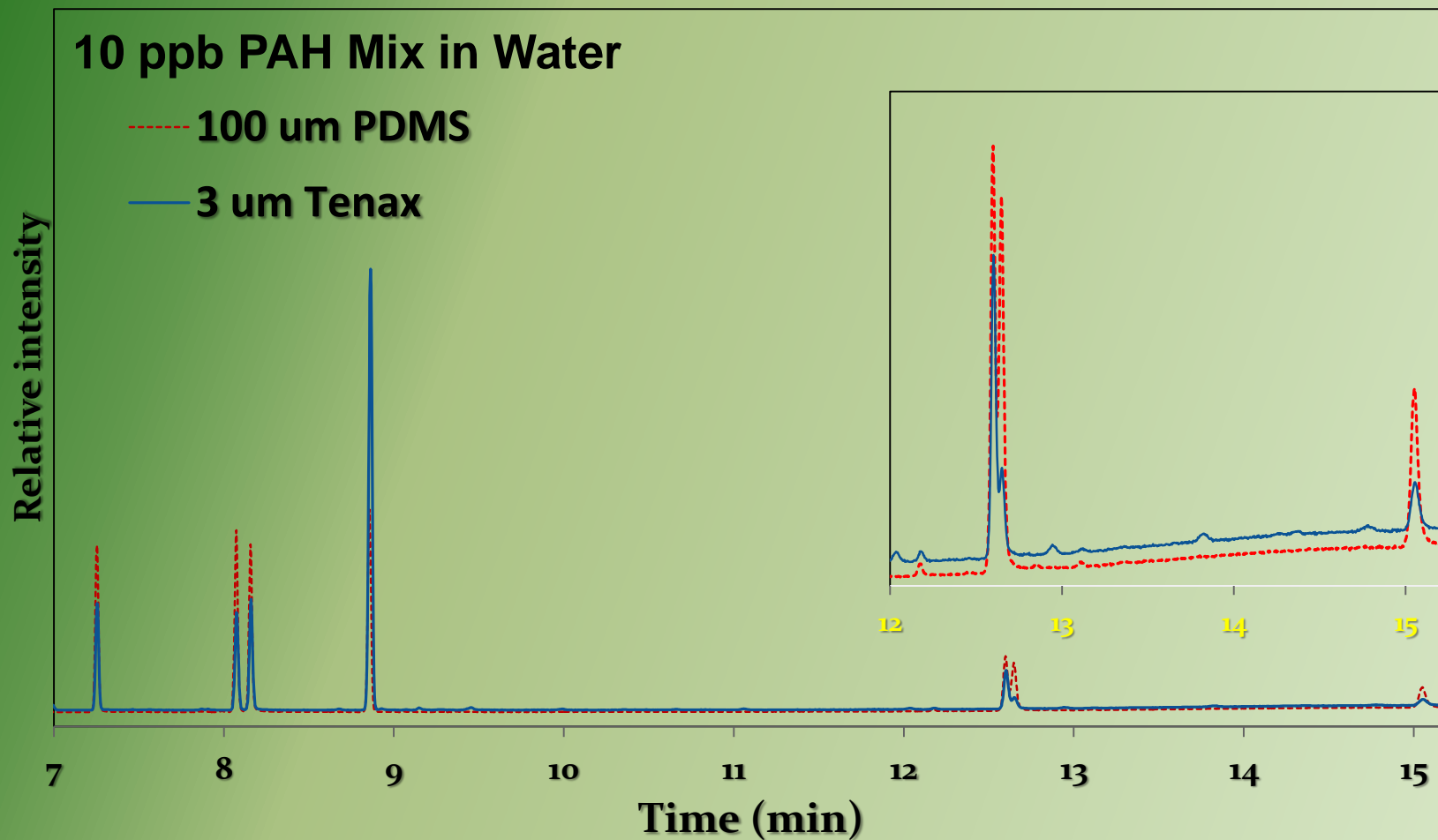
SPME – Solid Phase MicroExtraction



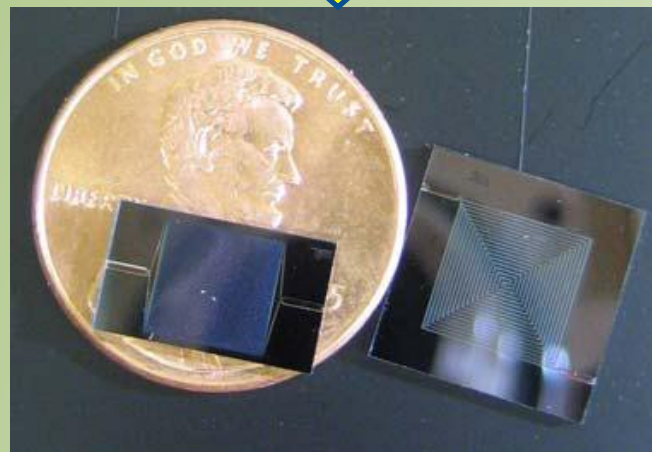
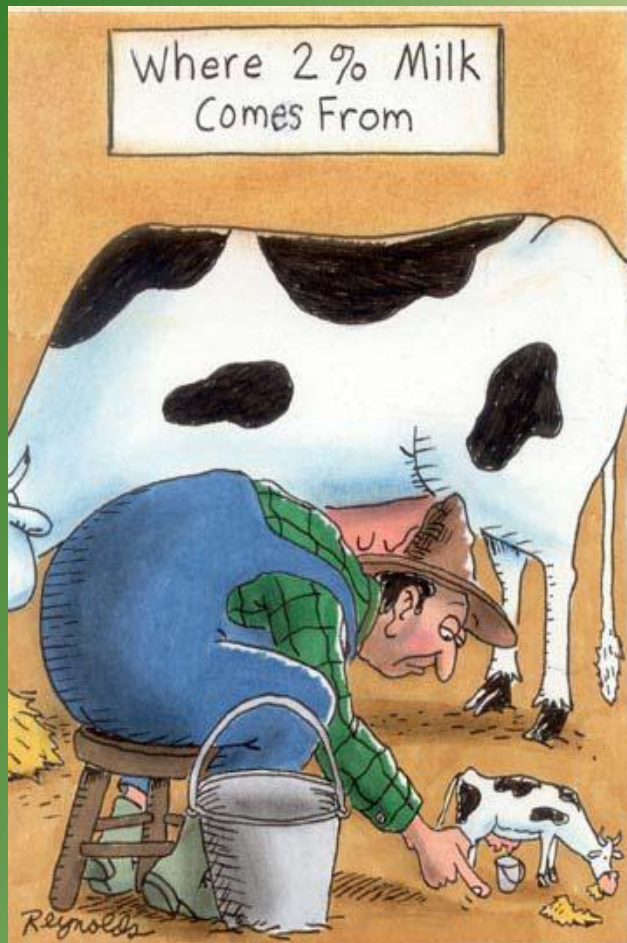
- Methodology for coating fibers with new adsorbent (Tenax) for high temperature applications ($>300\text{ }^{\circ}\text{C}$) being developed.
- Currently limited by 3 μm coating on the fiber. Needs to be 10-50 μm to compete with commercially available fibers.



SPME – Solid Phase MicroExtraction



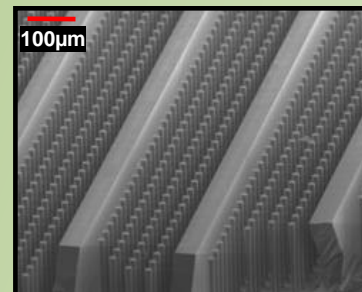
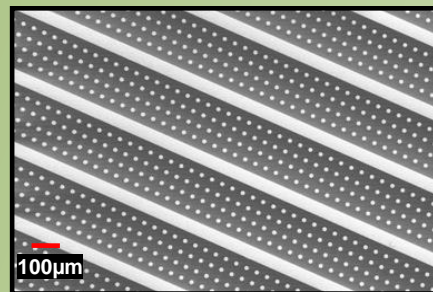
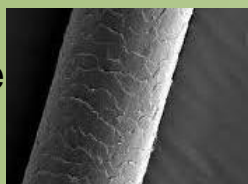
Micro-GC columns and Preconcentrators



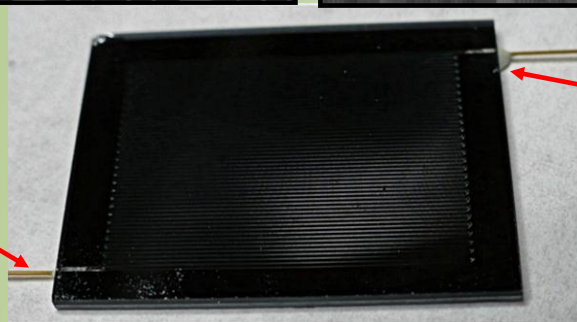
Micro-GC and Preconcentrators

- Significant interest in the development of low power micro-instrumentation
- Major obstacles in micro-GC include low sample capacity (for separations) and low breakthrough volumes (for preconcentration of analytes)
- One possible solution is to use columns consisting of nanofabricated columns throughout the primary column channel

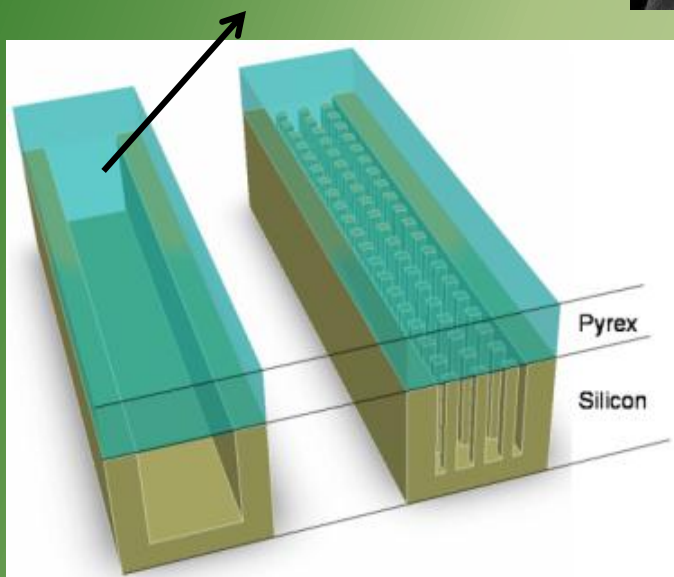
This width is comparable to the diameter of a human hair!!!



Fused Silica Capillary



Sealing Epoxy



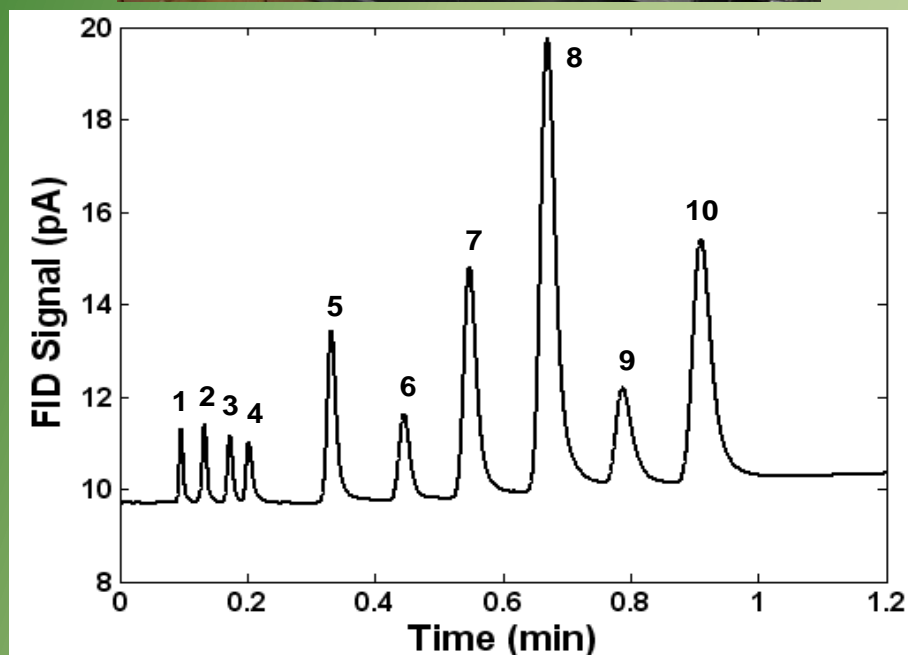
- Two NSF collaborations with Virginia Tech for which we are responsible for evaluating:
 - Column performance relative to conventional capillary columns for separations using different stationary phases, column coating techniques, and column architectures.
 - Performance of preconcentration adsorbers with respect to breakthrough volumes (sample capacity) using various stationary phases and column geometries.

MEMS (microelectromechanical systems) Column Evaluation



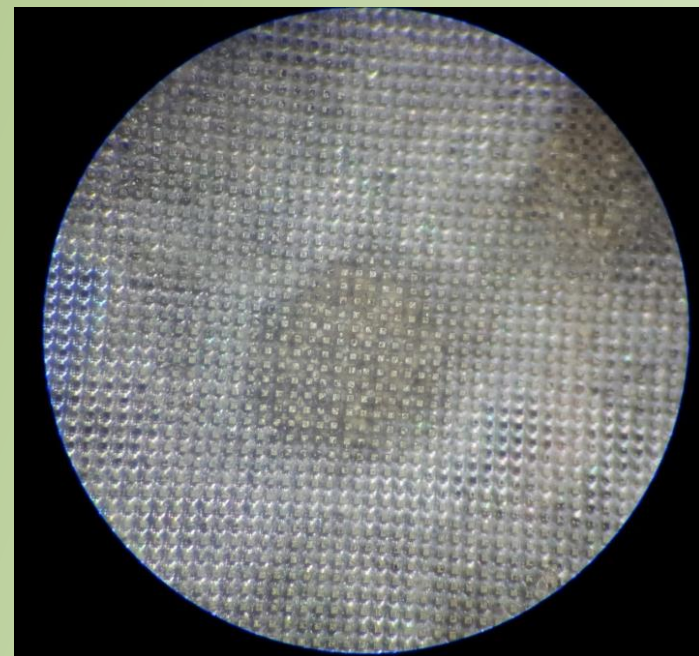
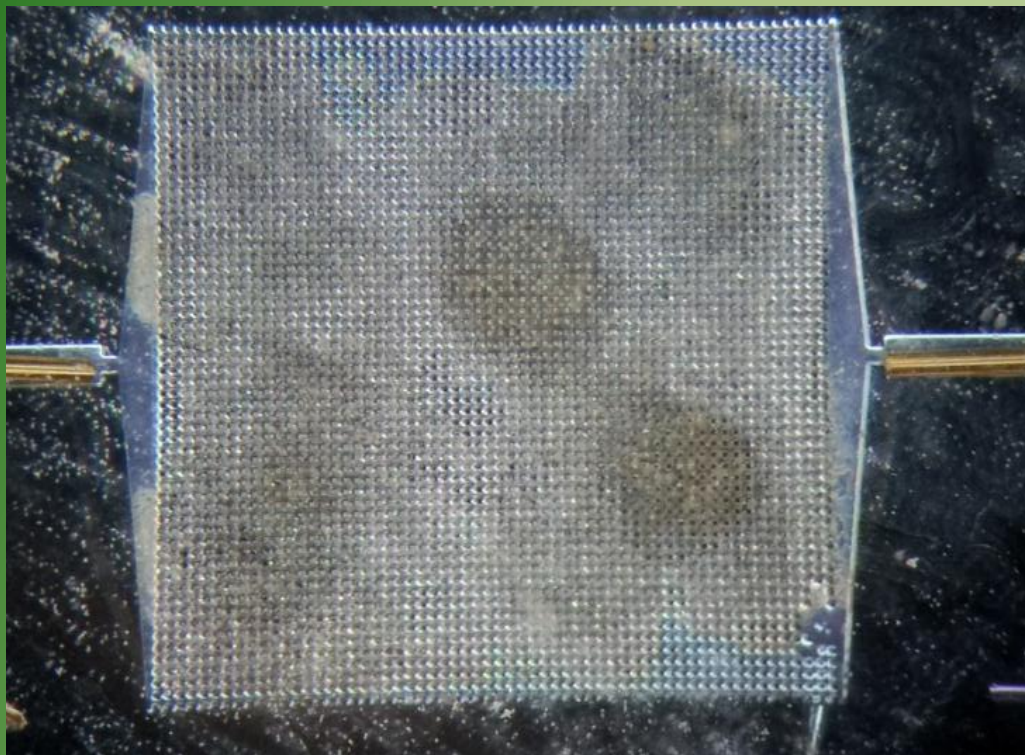
VOC vapor mix

- 1) Dichloromethane
- 2) Chloroform
- 3) Carbon tetrachloride
- 4) Dibromomethane
- 5) Toluene
- 6) Tetrachloroethylene
- 7) Chlorobenzene
- 8) p-xylene
- 9) 1,1,2,2-tetrachloroethane
- 10) Bromobenzene



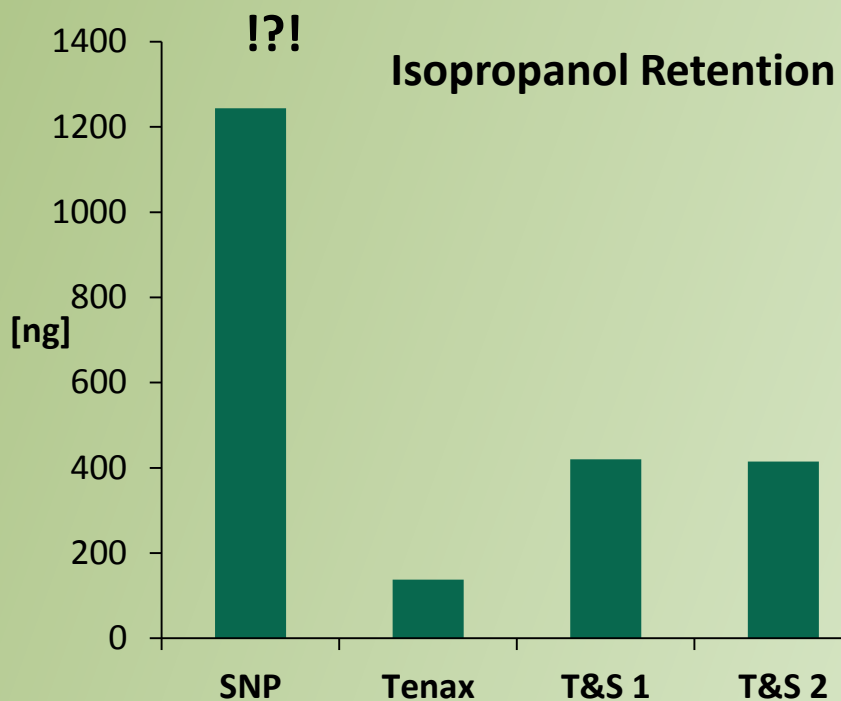
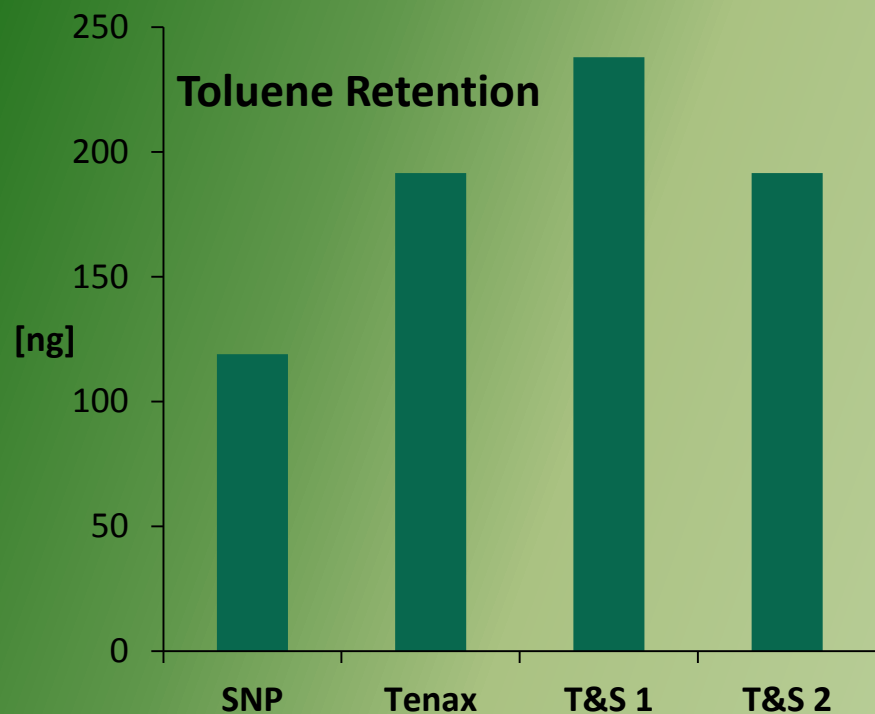
Micro-preconcentrators for micro-GC applications

What provides the best medium for maximum adsorption?



←→
~1 cm

Microconcentrator Performance



New column coating technology

In the **sol-gel** coating method, the surface treatment, deactivation, column coating, and immobilization occur simultaneously, not in individual steps.

Advantages:

Single step process greatly simplifies procedure

Polymer physically bonds to the column walls at active hydroxide sites

Process already available:

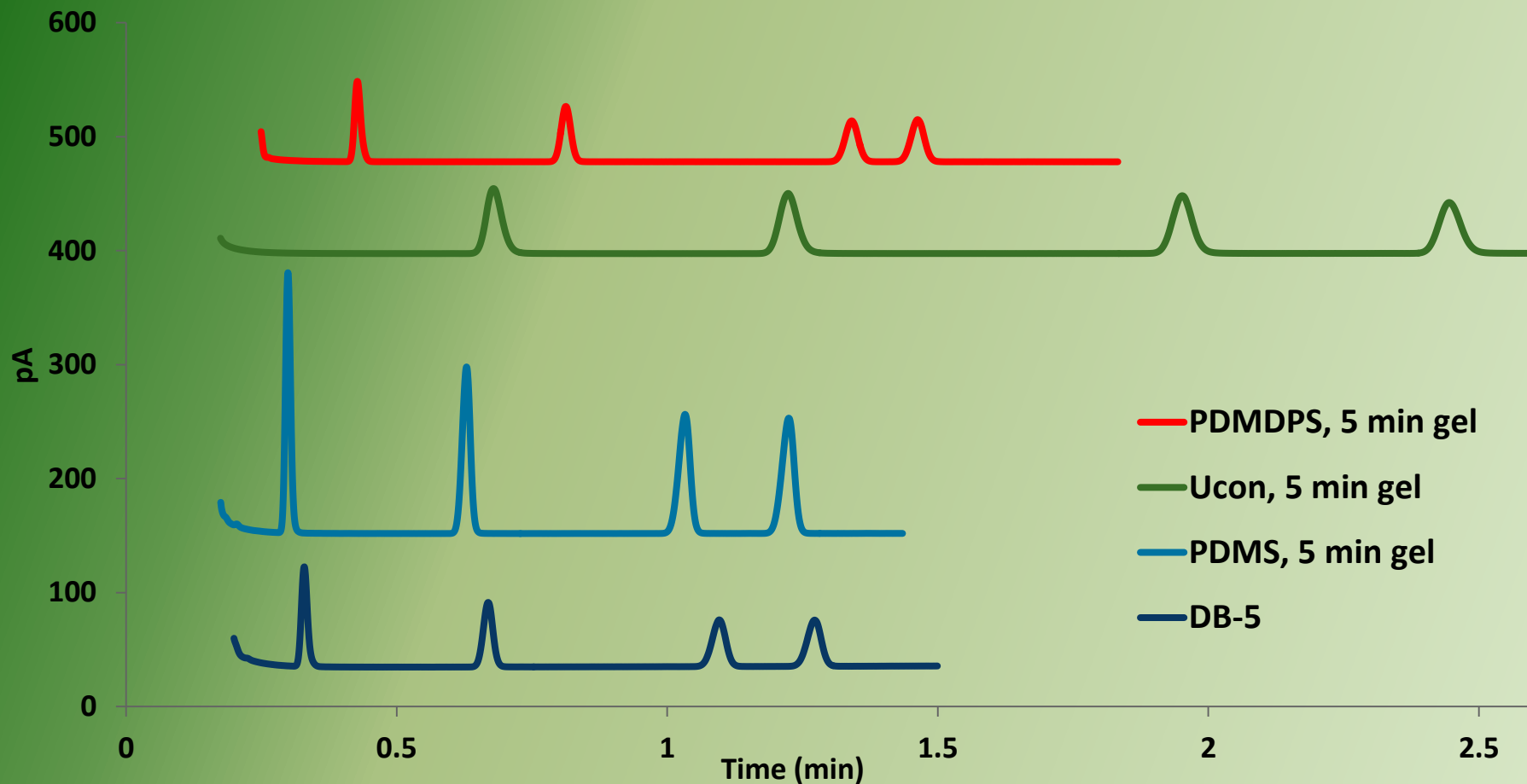
Sol-Gel Column Technology for Single-Step Deactivation, Coating, and Stationary-Phase Immobilization in High-Resolution Capillary Gas Chromatography. *Analytical Chem*, **1997**, 69, 4566-4576.

Malik, Abdul; Wang, Dongxin. Capillary Column and Method of Making. Patent No. WOO2000011463, March 2, 2000.



Materials	Function	Structure
poly(dimethylsiloxane) (PDMS)	stationary phase polymer	$\text{HO} - \left(\begin{array}{c} \text{CH}_3 \\ \\ \text{Si} - \text{O} \\ \\ \text{CH}_3 \end{array} \right)_n - \text{H}$
methyltrimethoxysilane (MTMS)	precursor	$\begin{array}{c} \text{CH}_3 \\ \\ \text{H}_3\text{C} - \text{O} - \text{Si} - \text{O} - \text{CH}_3 \\ \\ \text{O} \\ \\ \text{CH}_3 \end{array}$
poly(methylhydrosiloxane) (PMHS)	deactivation reagent	$\begin{array}{ccccccc} \text{CH}_3 & & \text{CH}_3 & & \text{CH}_3 & & \text{CH}_3 \\ & & & & & & \\ -\text{Si} & -\text{O}- & \text{Si} & -\text{O}- & \text{Si} & -\text{O}- & \text{Si}- \\ & & & & & & \\ \text{H} & & \text{CH}_3 & & \text{CH}_3 & & \text{H} \end{array}$
trifluoroacetic acid, 95% (TFA)	catalyst	CF_3COOH
dichloromethane	solvent	CH_2Cl_2

Separation of Aromatics



Test mix: benzene, toluene, chlorobenzene, xylene in pentane (1 ppthousand by volume)

Temperature: 30°C - 15°C/min - 80°C; Pressure: 5 psi

Stationary Phase	Sol-Gel Time (min)	Optimum Pressure (psi)	Theoretical Plates (per meter)
DB-5 (standard)	---	7.5	7080
PDMS sol-gel	5	5	5840
Ucon sol-gel	5	2.5	1590
PDMDPS sol-gel	5	5	5670
MEMS-SNP	10	5	2110

? QUESTIONS ?

